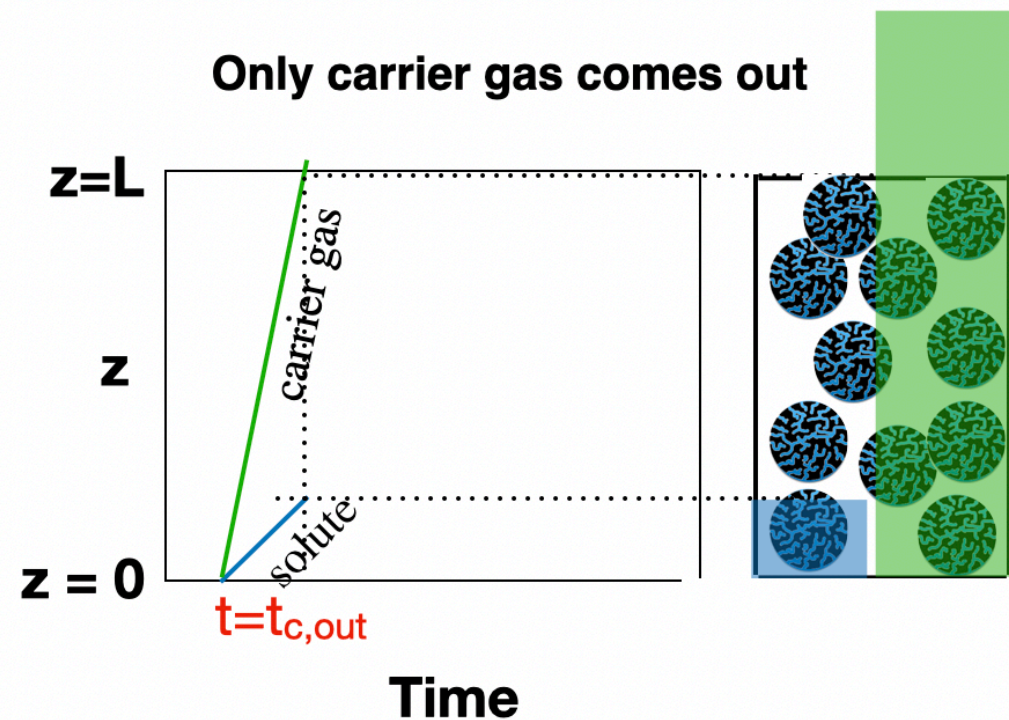


# Review quiz

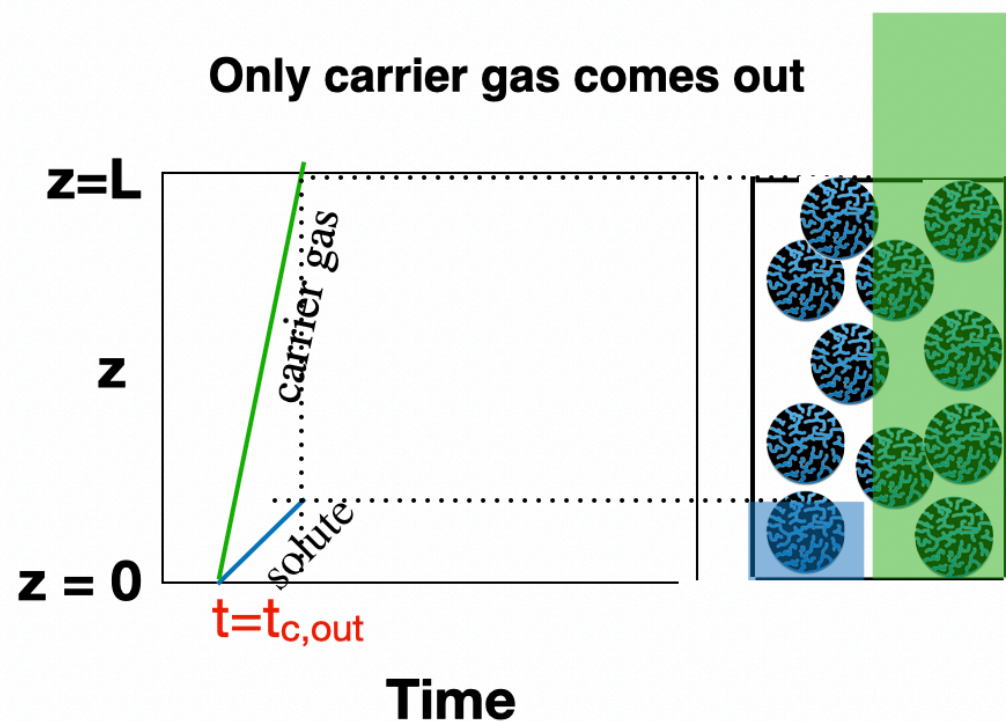
In the context of solute transport, which part has nearly zero velocity?



- A) Solute in inter-particle void
- B) Solute inside the particle
- C) Carrier gas
- D) Solute at column inlet

# Review quiz

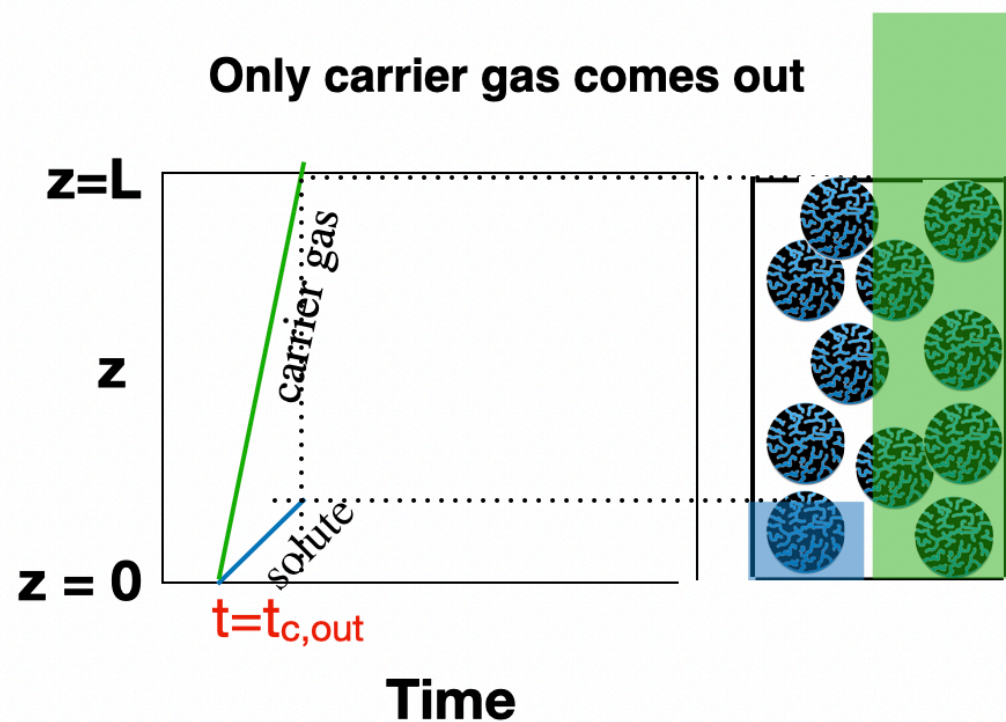
What is the purpose of reducing solute velocity in a packed column?



- A) To increase system pressure
- B) To allow for better separation
- C) To maximize temperature gradients
- D) Not correct. Solute velocity is same as carrier.

# Review quiz

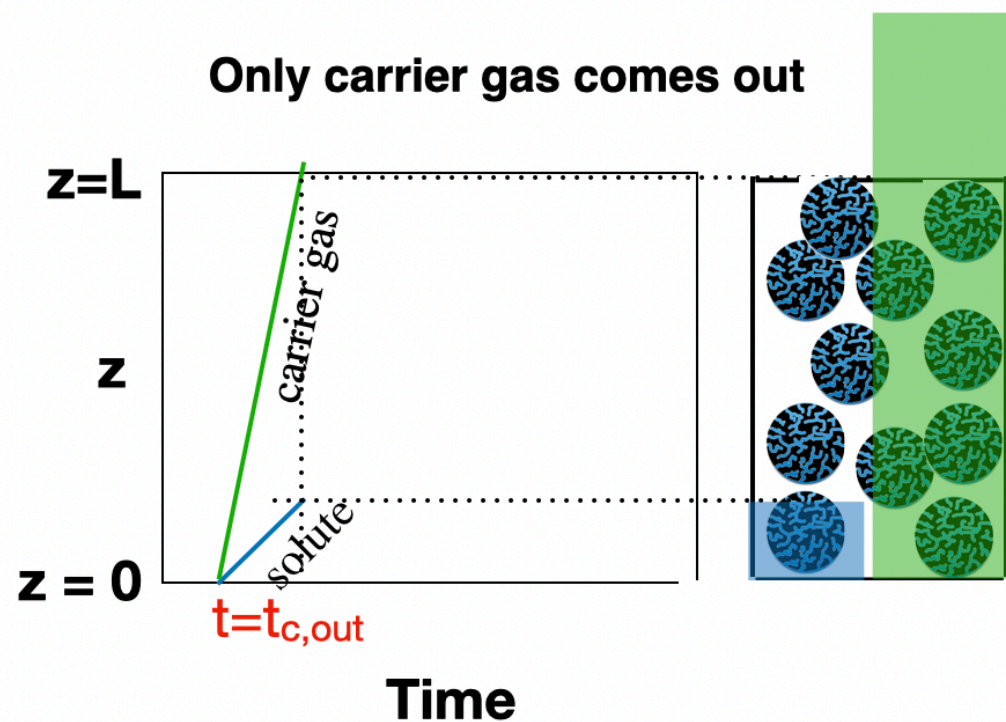
Why is the average velocity of solute in a packed column generally lower than the interstitial velocity?



- A) Solute molecules have a lower mass
- B) Most solute remains in the carrier gas phase
- C) Some solute diffuses into particles where velocity is negligible
- D) Interstitial velocity is affected by temperature only

# Review quiz

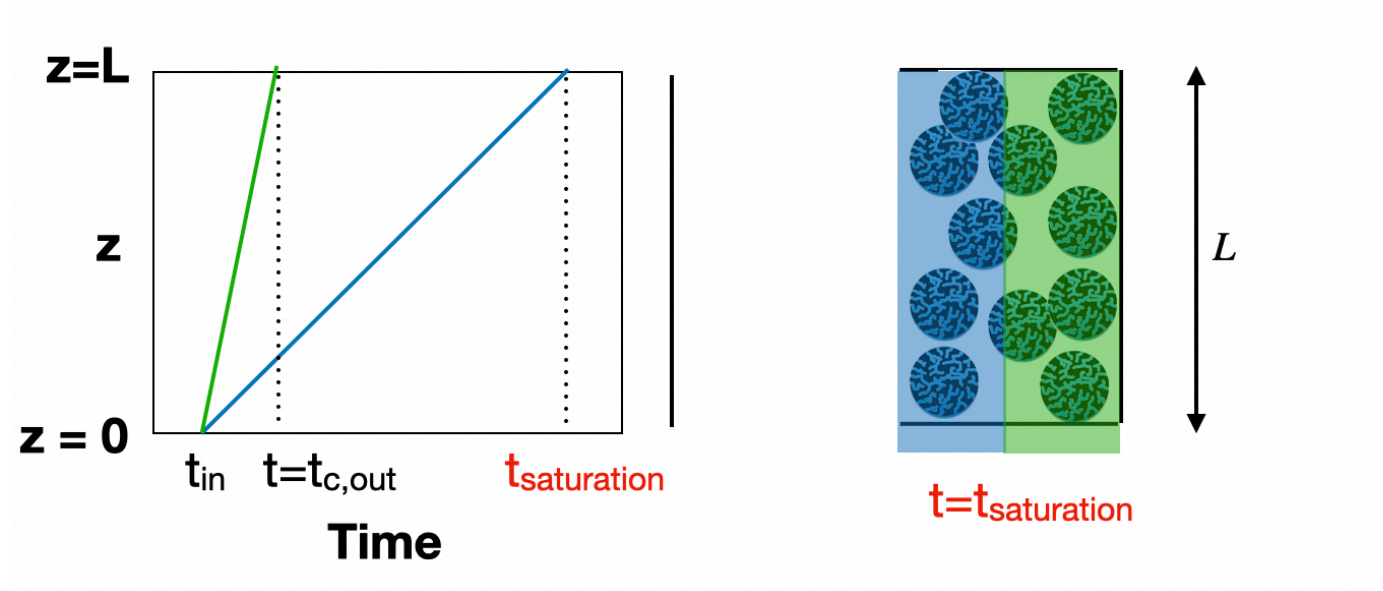
If the porosity  $\varepsilon_p$  of adsorbent pellets increases, what is the most likely impact on solute velocity?



- A) It will increase due to less adsorption
- B) It will decrease due to increased diffusive volume
- C) It will stay constant because porosity doesn't affect transport
- D) It will oscillate due to alternating flow patterns

# Review quiz

You have two trying to separate a mixture of helium and krypton in porous carbon packed bed. Which component will come out first



- A) Helium
- B) Krypton
- C) It depends on temperature of bed.
- D) It depends on velocity of mixture.



# In class exercise problem

You need to design an adsorption process to purify air from traces of hydrocarbon contaminants using a 2.0-meter-long adsorbent column. Air does not absorb at all.

- a) Calculate the interstitial velocity of solute.
- b) Calculate the average velocity of solute.
- c) Calculate the breakthrough time in minutes.
- d) How long can you use the adsorbent column in a single cycle for efficient separation.

$$\varepsilon_e = 0.50 \quad \varepsilon_p = 0.50 \quad K_d = 0.5 \quad \rho \frac{dq}{dc} = 10 \quad A_c = 1 \text{ m}^2$$

Amount of hydrocarbon in feed (inlet) air = 1 %

Feed air flow rate = 1 m<sup>3</sup> /minute

# In class exercise problem

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Amount of hydrocarbon in feed (inlet) air = 1 %

Feed air flow rate = 1 m<sup>3</sup> /minute

$$\text{Flowrate of hydrocarbon} = Q_h = 1 * 0.01 \text{ m}^3/\text{minute}$$

$$v_{inter} = \frac{Q_h}{A_c \varepsilon_e} = \frac{0.01}{1 * 0.5} = 0.02 \text{ m/min}$$

$$u_s = \frac{v_{inter}}{1 + \left( \frac{1 - \varepsilon_e}{\varepsilon_e} \right) * (K_d \varepsilon_p) + \left( \frac{1 - \varepsilon_e}{\varepsilon_e} \right) * (1 - \varepsilon_p) * \rho \left( \frac{\Delta q}{\Delta c} \right)}$$

$$\Rightarrow u_s = \frac{0.02}{1 + 0.25 + 0.5 * 10} = 0.0032 \text{ m/minute}$$

$$\text{Breakthrough time} = t_{saturation} - t_{in} = \frac{L}{u_s} = 625 \text{ minutes}$$